

Graduate Aptitude Test in Engineering 2017

Question Paper Name: Life Sciences 12th Feb 2017
Subject Name: Life Sciences
Duration: 180
Total Marks: 100



Organizing Institute:
Indian Institute of Technology Roorkee



Chemistry (XL-P) (Compulsory)

Question Number : 1

Correct : 1 Wrong : -0.33

CO reacts readily with

- (A) Fe (B) Fe^{2+} (C) Fe^{4+} (D) Fe^{3+}

Question Number : 2

Correct : 1 Wrong : -0.33

Molecules that are **NOT** isoelectronic to NO_2^+ ion are

- (A) CO_2 and N_3^-
(B) NCO^- and H_3BCN^-
(C) BO_2^- and $\text{H}_3\text{CC}\equiv\text{CH}$
(D) OF_2 and O_3^-

Question Number : 3

Correct : 1 Wrong : -0.33

The extensive quantity among the following is

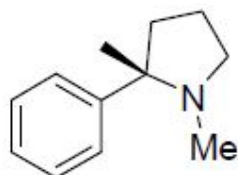
- (A) Pressure
(B) Temperature
(C) Chemical potential
(D) Volume

Question Number : 4

Correct : 1 Wrong : -0.33

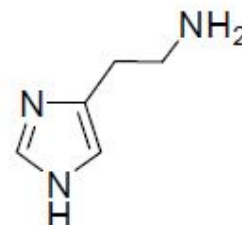
The compound that gives characteristic foul smell upon heating with potassium hydroxide and chloroform is

(A)



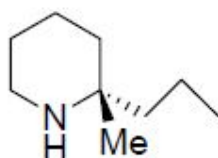
Nicotine

(B)



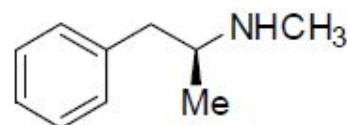
Histamine

(C)



Coniine

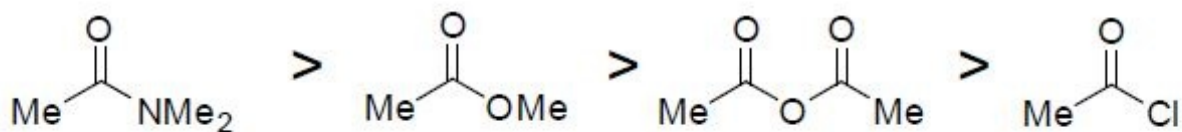
(D)



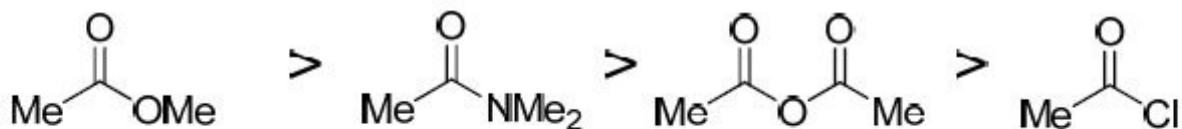
Methamphetamine

The correct order of stability in water is

(A)



(B)



(C)



(D)



The pair of molecules having non-linear structures is

(A) ICl_2^- and BeH_2

(B) CS_2 and I_3^-

(C) SCl_2 and ClO_2^-

(D) XeF_2 and CN_2^{2-}

Question Number : 7

Correct : 2 Wrong : -0.66

The decreasing order of bond lengths for O_2 , B_2 , N_2 and C_2 is

- (A) $B_2 > C_2 > N_2 > O_2$
- (B) $B_2 > C_2 > O_2 > N_2$
- (C) $N_2 > C_2 > O_2 > B_2$
- (D) $B_2 > O_2 > N_2 > C_2$

Question Number : 8

Correct : 2 Wrong : -0.66

The octahedral metal oxide with the highest CFSE value is

- (A) ZnO
- (B) MnO
- (C) VO
- (D) TiO

Question Number : 9

Correct : 2 Wrong : -0.66

Assuming independent non-interacting electrons, the first ionization energy of Helium atom is

- (A) 13.6 eV
- (B) 27.2 eV
- (C) 54.4 eV
- (D) 108.8 eV

Question Number : 10

Correct : 2 Wrong : -0.66

For a reaction $A + B \longrightarrow$ products, the following data was obtained.

$[A]_0$ (M)	$[B]_0$ (M)	Initial rate
0.1	0.1	r
0.2	0.1	4r
0.1	0.2	2r

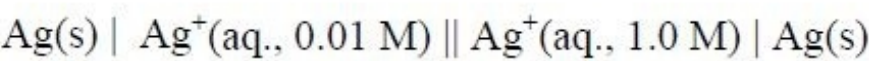
A_0 and B_0 are initial concentrations of A and B, respectively. The overall order of the reaction is

- (A) 2
- (B) 3
- (C) 4
- (D) 6

Question Number : 11

Correct : 2 Wrong : -0.66

The EMF for the following cell at 298.15 K is



(Standard reduction potential for $\text{Ag}^+ + \text{e}^- \longrightarrow \text{Ag}$ is -0.80 V)

- (A) 0.12 V (B) 0.68 V (C) 0.80 V (D) 0.92 V

Question Number : 12

Correct : 2 Wrong : 0

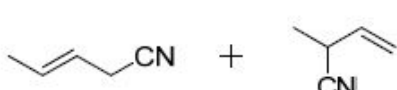
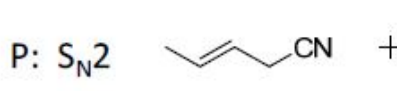
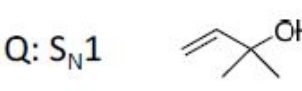
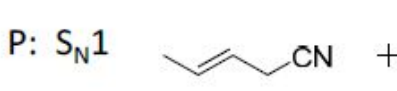
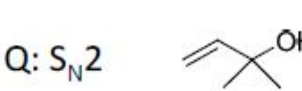
One gram of a protein is dissolved in one liter of water. The resulting solution exerts an osmotic pressure of 1.4 Torr at 298 K . Assuming that the protein does not ionize in solution, the molecular weight of the protein is _____ g mol^{-1} . ($R = 0.082 \text{ L atm mol}^{-1} \text{ K}^{-1}$)

Question Number : 13

Correct : 2 Wrong : -0.66

The type of nucleophilic substitution and the possible products for each of the reactions P and Q are

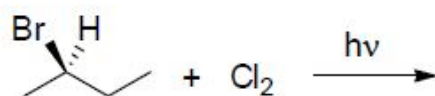


- (A) P: $\text{S}_{\text{N}}2$  Q: $\text{S}_{\text{N}}1$ 
- (B) P: $\text{S}_{\text{N}}1$  Q: $\text{S}_{\text{N}}2$ 
- (C) P: $\text{S}_{\text{N}}2$  Q: $\text{S}_{\text{N}}1$ 
- (D) P: $\text{S}_{\text{N}}1$  Q: $\text{S}_{\text{N}}2$ 

Question Number : 14

Correct : 2 Wrong : -0.66

If mono-chlorination occurs at every carbon in the following reaction, the number of isomers (stereo isomers + constitutional isomers) that one can have is

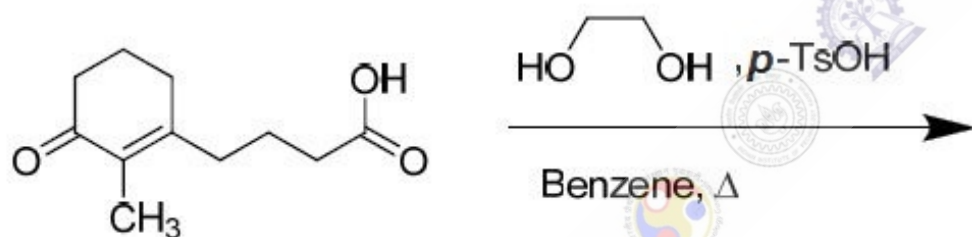


- (A) 4 (B) 5 (C) 6 (D) 8

Question Number : 15

Correct : 2 Wrong : -0.66

The major product in the following reaction is



- (A)
- (B)
- (C)
- (D)

General Aptitude

Question Number : 116

Correct : 1 Wrong : -0.33

The event would have been successful if you _____ able to come.

- (A) are (B) had been (C) have been (D) would have been

Question Number : 117

Correct : 1 Wrong : -0.33

There was no doubt that their work was thorough.

Which of the words below is closest in meaning to the underlined word above?

- (A) pretty (B) complete (C) sloppy (D) haphazard

Question Number : 118

Correct : 1 Wrong : -0.33

Four cards lie on a table. Each card has a number printed on one side and a colour on the other. The faces visible on the cards are 2, 3, red, and blue.

Proposition: If a card has an even value on one side, then its opposite face is red.

The cards which MUST be turned over to verify the above proposition are

- (A) 2, red (B) 2, 3, red (C) 2, blue (D) 2, red, blue

Question Number : 119

Correct : 1 Wrong : -0.33

What is the value of x when $81 \times \left(\frac{16}{25}\right)^{x+2} \div \left(\frac{3}{5}\right)^{2x+4} = 144$?

- (A) 1 (B) -1 (C) -2 (D) Cannot be determined

Question Number : 120

Correct : 1 Wrong : -0.33

Two dice are thrown simultaneously. The probability that the product of the numbers appearing on the top faces of the dice is a perfect square is

- (A) $\frac{1}{9}$ (B) $\frac{2}{9}$ (C) $\frac{1}{3}$ (D) $\frac{4}{9}$

Question Number : 121

Correct : 2 Wrong : -0.66

Bhaichung was observing the pattern of people entering and leaving a car service centre. There was a single window where customers were being served. He saw that people inevitably came out of the centre in the order that they went in. However, the time they spent inside seemed to vary a lot: some people came out in a matter of minutes while for others it took much longer.

From this, what can one conclude?

- (A) The centre operates on a first-come-first-served basis, but with variable service times, depending on specific customer needs.
(B) Customers were served in an arbitrary order, since they took varying amounts of time for service completion in the centre.
(C) Since some people came out within a few minutes of entering the centre, the system is likely to operate on a last-come-first-served basis.
(D) Entering the centre early ensured that one would have shorter service times and most people attempted to do this.

Question Number : 122

Correct : 2 Wrong : -0.66

A map shows the elevations of Darjeeling, Gangtok, Kalimpong, Pelling, and Siliguri. Kalimpong is at a lower elevation than Gangtok. Pelling is at a lower elevation than Gangtok. Pelling is at a higher elevation than Siliguri. Darjeeling is at a higher elevation than Gangtok.

Which of the following statements can be inferred from the paragraph above?

- i. Pelling is at a higher elevation than Kalimpong
ii. Kalimpong is at a lower elevation than Darjeeling
iii. Kalimpong is at a higher elevation than Siliguri
iv. Siliguri is at a lower elevation than Gangtok
- (A) Only ii (B) Only ii and iii (C) Only ii and iv (D) Only iii and iv

Correct : 2 Wrong : -0.66

(A) P is immediately to the right of R
(B) T is immediately to the left of P
(C) T is immediately to the left of P or P is immediately to the right of Q
(D) U is immediately to the right of R or P is immediately to the left of T

Correct : 2 Wrong : -0.66

(A) 1 (B) 4 (C) 5 (D) 6

Correct : 2 Wrong : -0.66

- i. The elevator never moves directly from any non-ground floor to another non-ground floor over the one hour period
- ii. The elevator stays on the fourth floor for the longest duration over the one hour period

- (A) Only i (B) Only ii (C) Both i and ii (D) Neither i nor ii

Q. No.	Type	Section	Key	Marks
1	MCQ	XL-P	A	1
2	MCQ	XL-P	D	1
3	MCQ	XL-P	D	1
4	MCQ	XL-P	B	1
5	MCQ	XL-P	A	1
6	MCQ	XL-P	C	2
7	MCQ	XL-P	B	2
8	MCQ	XL-P	C	2
9	MCQ	XL-P	C	2
10	MCQ	XL-P	B	2
11	MCQ	XL-P	A	2
12	NAT	XL-P	13260 to 13285	2
13	MCQ	XL-P	A	2
14	MCQ	XL-P	C	2
15	MCQ	XL-P	A	2
16	MCQ	XL-Q	A	1
17	MCQ	XL-Q	D	1
18	MCQ	XL-Q	C	1
19	NAT	XL-Q	3.0 to 3.0	1
20	MCQ	XL-Q	B	1
21	MCQ	XL-Q	D	1
22	MCQ	XL-Q	C	1
23	MCQ	XL-Q	A	1
24	MCQ	XL-Q	C	1
25	MCQ	XL-Q	D	1
26	NAT	XL-Q	7575 to 7576	2
27	NAT	XL-Q	722 to 725	2
28	NAT	XL-Q	0.19 to 0.20	2
29	MCQ	XL-Q	D	2
30	NAT	XL-Q	9.0 to 9.0	2
31	MCQ	XL-Q	B	2
32	MCQ	XL-Q	A	2
33	MCQ	XL-Q	A	2
34	MCQ	XL-Q	A	2
35	MCQ	XL-Q	B	2
36	MCQ	XL-R	C	1

37	MCQ	XL-R	B	1
38	MCQ	XL-R	C	1
39	MCQ	XL-R	A	1
40	MCQ	XL-R	D	1
41	MCQ	XL-R	A	1
42	MCQ	XL-R	C	1
43	MCQ	XL-R	D	1
44	MCQ	XL-R	D	1
45	MCQ	XL-R	A	1
46	MCQ	XL-R	A	2
47	MCQ	XL-R	A	2
48	MCQ	XL-R	B	2
49	NAT	XL-R	18.3 to 18.5	2
50	NAT	XL-R	3.2 to 3.2	2
51	MCQ	XL-R	B	2
52	MCQ	XL-R	D	2
53	MCQ	XL-R	C	2
54	MCQ	XL-R	D	2
55	MCQ	XL-R	A	2
56	MCQ	XL-S	A	1
57	MCQ	XL-S	B	1
58	MCQ	XL-S	D	1
59	MCQ	XL-S	C	1
60	MCQ	XL-S	C	1
61	MCQ	XL-S	A	1
62	MCQ	XL-S	A	1
63	MCQ	XL-S	D	1
64	MCQ	XL-S	B	1
65	MCQ	XL-S	C	1
66	MCQ	XL-S	A	2
67	MCQ	XL-S	B	2
68	MCQ	XL-S	A	2
69	MCQ	XL-S	B	2
70	MCQ	XL-S	B	2
71	NAT	XL-S	4550000 to 4650000	2
72	NAT	XL-S	99 to 101	2
73	NAT	XL-S	78000 to 82000	2

74	NAT	XL-S	9990 to 10020	2
75	MCQ	XL-S	A	2
76	MCQ	XL-T	D	1
77	MCQ	XL-T	A	1
78	MCQ	XL-T	C	1
79	MCQ	XL-T	B	1
80	MCQ	XL-T	C	1
81	MCQ	XL-T	A	1
82	MCQ	XL-T	C	1
83	MCQ	XL-T	B	1
84	MCQ	XL-T	D	1
85	MCQ	XL-T	D	1
86	NAT	XL-T	8.0 to 8.0	2
87	NAT	XL-T	7.0 to 7.0	2
88	NAT	XL-T	0.55 to 0.55	2
89	NAT	XL-T	16 to 16	2
90	MCQ	XL-T	D	2
91	MCQ	XL-T	A	2
92	MCQ	XL-T	B or D	2
93	MCQ	XL-T	B	2
94	NAT	XL-T	42.0 to 42.0	2
95	NAT	XL-T	250 to 250	2
96	MCQ	XL-U	B	1
97	MCQ	XL-U	B	1
98	MCQ	XL-U	D	1
99	MCQ	XL-U	B	1
100	NAT	XL-U	1000 to 1000	1
101	MCQ	XL-U	A	1
102	NAT	XL-U	2078 to 2086	1
103	NAT	XL-U	0.80 to 0.90	1
104	MCQ	XL-U	A	1
105	MCQ	XL-U	D	1
106	MCQ	XL-U	C	2
107	MCQ	XL-U	B	2
108	MCQ	XL-U	D	2
109	NAT	XL-U	25550 to 25650	2
110	MCQ	XL-U	D	2

111	NAT	XL-U	53.30 to 55.25	2
112	NAT	XL-U	64.0 to 64.0	2
113	NAT	XL-U	10.80 to 11.20	2
114	NAT	XL-U	9000 to 9000	2
115	NAT	XL-U	0.30 to 0.40	2
116	MCQ	GA	B	1
117	MCQ	GA	B	1
118	MCQ	GA	C	1
119	MCQ	GA	B	1
120	MCQ	GA	B	1
121	MCQ	GA	A	2
122	MCQ	GA	C	2
123	MCQ	GA	C	2
124	MCQ	GA	D	2
125	MCQ	GA	D	2